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**ЦЕНТР НАУКОВИХ ПУБЛІКАЦІЙ
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ТЕХНІЧНІ НАУКИ

Cu(II)–(ПОЛІАМІН) ХЕЛАТУВАННЯ ЯК ЕФЕКТИВНИЙ СПОСІБ ЗНИЖЕННЯ ГОРЮЧОСТІ ЕПОКСІ-АМІННИХ КОМПОЗИЦІЙ МОДИФІКОВАНИХ НЕОРГАНІЧНИМИ СОЛЯМИ КУПРУМУ(II)

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Cu(II)–(POLYAMINE) CHELATION AS AN EFFICIENT WAY OF REDUCING COMBUSTIBILITY OF EPOXY-AMINE COMPOSITES MODIFIED BY COPPER(II) INORGANIC SALTS

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Анотація

Було розроблено технологію виготовлення купрум(II)-координованих епоксі-амінних композитів зі зниженою горючістю, що дозволило сформувати основу нового типу полімеру з підвищеними фізико-хімічними та механічними властивостями. Був запропонований простий спосіб, виготовлення антипіренів-затверджувачів та включення їх в матрицю епоксидної смоли. Останні є продуктами взаємодії поліетиленуполіаміну (*pepa*), що містить етилендіамін (*eda*) та діетилентріамін (*deta*) з неорганічними солями купруму(II) (зокрема CuSO_4 , $(\text{CuOH})_2\text{CO}_3$ та CuSiF_6). Горючість отриманих композицій *DGEBA/pepa*, *DGEBA/deta-CuSO₄*, *DGEBA/deta-CuCO₃* і *DGEBA/pepa-CuSiF₆* (*DGEBA* – це дигліцидиловий ефір бісфенолу А) досліджували методом керамічної труби. Займистість отриманих зразків оцінювалась за допомогою тестів UL94. Описано вплив неорганічних солей купруму(II) на пожежонебезпечні показники модифікованих епоксі-амінних композитів.

Abstract

Making technology of copper(II)-coordinated epoxy-amine composites with reduced combustibility has been developed that allowed to form the basis of a new type of polymer with enhanced physicochemical and mechanical properties. A proposed simple method allows preparing and incorporating into the epoxy resin matrix the fire retardant-hardeners. These are considered to be the interaction products of the polyethylenepolyamine (*pepa*) containing ethylenediamine (*eda*) and diethylenetriamine (*deta*) with inorganic copper(II) salts (specifically,

CuSO₄, (CuOH)₂CO₃ and CuSiF₆). The combustibility of the DGEBA/pepa, DGEBA/deta-CuSO₄, DGEBA/deta-CuCO₃ and DGEBA/pepa-CuSiF₆ composites obtained (DGEBA is bisphenol A diglycidyl ether) were investigated using ceramic tube method. The flammability of the polymer samples has been evaluated using the UL94 tests. The effects of copper(II) inorganic salts on the fire-hazardous indices of the modified epoxy-amine composites were described.

Ключові слова: купрум(II)-координовані епоксі-амінні композити, антипірен-затвердник, самозгасаючий характер горіння, понижена горючість.

Keywords: copper(II)-coordinated epoxy-amine composites, fire retardant-hardener, self-extinguishing character of burning, reduced combustibility

Polyethylenepolyamine (*pepa*) containing ethylenediamine (*eda*) and diethylenetriamine (*deta*) is known to widely be used as a hardener at the polymerization of epoxy resins. On the other hand, *pepa* (*eda* + *deta*) is a potentially tridentate ligand which can chelate atoms of transition metals, in particular, Cu(II). This behavior feature of *pepa* has allowed us to solve a problem elaborating a new type of polymer materials with reduced combustibility using, at that, the following approach. The flame retardant agent *viz.* CuSO₄ [1], (CuOH)₂CO₃ [2] and CuSiF₆ [3] was added to the epoxy-amine polymer. In this process the chemical interaction between Cu(II) atoms of said inorganic salts and *pepa* (*eda* + *deta*) molecules predetermines lowering combustibility of resulting epoxy-amine composites [4–6]. This can be explained to form virtually non-combustible chelate complexes – [Cu(*deta*)H₂O]SO₄·H₂O, [Cu(*deta*)(H₂O)(CO₃)₂]·6H₂O, and [Cu(*deta*)(*eda*)]SiF₆ [7–9].

The *ab initio* quantum-chemical calculations of charge density distribution on atoms and the bonding energy carried out for *pepa* in coordinated and uncoordinated state reveal that Cu(II)–(*pepa*) chelation is accompanied by some changes in the electronic parameters for coordinated *pepa* in comparison with uncoordinated *pepa* [10–14]. So, the electron density of all N atoms within coordination chelate-core efficiently shifts to the central atom of Cu(II) due to chelation effect. At that, a lone-electron pair of each N atom of amino-groups more efficiently overlap with the four of six unoccupied hybrid $sp^3d_{x^2-y^2}d_{z^2}$ AOs of the central Cu(II) atom. At the same time, the total energy of three Cu–N bonds which belong to the square-pyramidal coordination core is 237.39 kJ·mol⁻¹. The initial octahedral surrounding of the Cu(II) atom is strongly distorted and acquires the shape of the extended square pyramid. As a result, the square-pyramidal field of the ligands simultaneously with chelate effect eliminates the degeneracy of the 3*d*-AOs of Cu²⁺ ion. Thus, the geometrical parameters of the square-pyramidal polyhedron of Cu(II) atom are in good agreement with the Jahn-Teller principle and are the result of the mutual strengthening of two individual effects (of square-pyramidal crystal field effect and chelate effect). This synergism causes the splitting of the

degenerate $3d$ -AOs of the Cu^{2+} ion onto four sets of the energy levels ($d_{xy} <$ twice degenerate d_{xz} and $d_{yz} < d_{z^2} < d_{x^2-y^2}$).

In summary, the strong coordination bonds that arise between Cu(II) atoms of incombustible inorganic salts and N atoms of the amine hardener as well as formation of stable chelate complexes in solid state are responsible for the combustibility suppression of the epoxy-amine composites modified by copper(II) inorganic salts.

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